

Cu-Fe-Se (Copper-Iron-Selenium)

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The previous review of this system by [1992Rag] presented isothermal sections at 900, 700, and 500 °C, from the results of Bernardini et al. [1979Ber, 1981Ber]. Based on the isothermal sections, a schematic liquidus projection and a reaction scheme were also included in the evaluation by [1992Rag]. Koneshova [1991Kon] determined a pseudobinary section of this system along the $\text{Cu}_2\text{Se-FeSe}$ join.

Binary Systems

The Cu-Fe phase diagram [1993Swa] shows that there are no intermediate phases in this system. A metastable liquid miscibility gap is known in this system. The Cu-Se phase diagram [1981Cha] depicts four intermediate phases: Cu_{2-x}Se , Cu_3Se_2 , CuSe , and CuSe_2 . Among these, Cu_{2-x}Se has two modifications (monoclinic and cubic). The high temperature cubic structure, $\beta\text{Cu}_{2-x}\text{Se}$, forms congruently at 1130 °C. The monoclinic $\alpha\text{Cu}_{2-x}\text{Se}$ forms peritectoidally at 123 °C. CuSe has three crystalline forms: α (hexagonal), β

(orthorhombic), and γ (hexagonal). Cu_3Se_2 is tetragonal and CuSe_2 has the FeS_2 (marcasite) type orthorhombic structure. The Fe-Se phase diagram [1991Oka] depicts a number of modifications of the monoselenide around the midcomposition: $\beta\text{Fe}_{1.04}\text{Se}$, $\gamma\text{Fe}_{1-x}\text{Se}$, $\gamma'\text{Fe}_{1-x}\text{Se}$, $\delta\text{Fe}_{1-x}\text{Se}$, and $\delta'\text{Fe}_{1-x}\text{Se}$. $\beta\text{Fe}_{1.04}\text{Se}$ has the tetragonal PbO type structure. $\delta\text{Fe}_{1-x}\text{Se}$ is a NiAs-type hexagonal phase. The other phases are NiAs-related phases. FeSe_2 has the FeS_2 (marcasite) type orthorhombic structure. For more structural details, see [1991Oka].

Ternary Compounds

As summarized by [1992Rag], there are two ternary compounds in this system with significant homogeneity ranges. $(\text{Cu,Fe})\text{Se}_{2-x}$, with the cubic pyrite (FeS_2) type structure, has the mineral name eskebornite. A high-pressure phase transition in eskebornite was reported by [1996Tin]. $(\text{Cu,Fe})\text{Se}$ occurs approximately midway between the Cu and Fe monoselenides and has the Cu_2Sb type

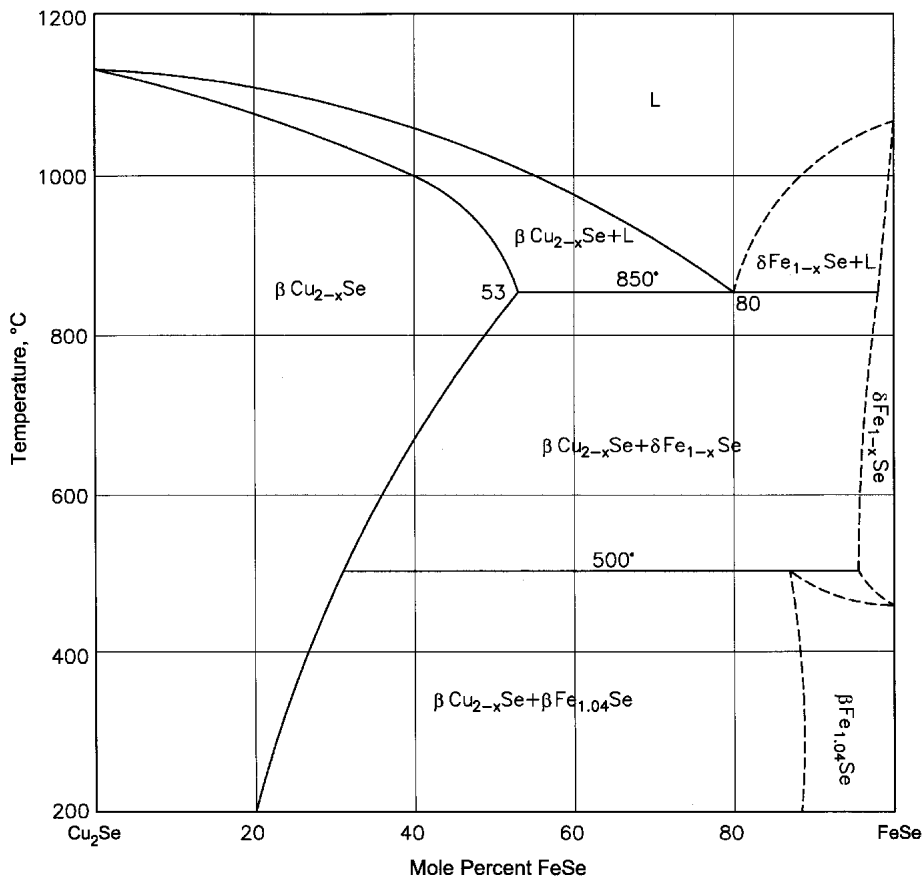


Fig. 1 Cu-Fe-Se pseudobinary section along the $\text{Cu}_2\text{Se-FeSe}$ join [1991Kon]

Section II: Phase Diagram Evaluations

tetragonal structure. Neither of these is present along the Cu_2Se -FeSe join [1991Kon].

The FeSe- Cu_2Se Pseudobinary Section

With high-purity starting materials, [1991Kon] synthesized about ten alloy compositions in evacuated quartz tubes at 950 °C, which were then annealed at 475 °C for 30 d. The phase equilibria were studied by differential thermal analysis, x-ray diffraction, metallography, and microhardness measurements. The pseudobinary section along the Cu_2Se -FeSe join constructed by [1991Kon] is redrawn in Fig. 1 to agree with accepted binary data. There is a eutectic reaction at 850 °C and at 80 mol% FeSe, which yields $\delta\text{Fe}_{1-x}\text{Se}$ and $\beta\text{Cu}_{2-x}\text{Se}$ containing up to 53 mol% FeSe. The solubility of FeSe in $\beta\text{Cu}_{2-x}\text{Se}$ decreases rapidly with decreasing temperature. A peritectoid reaction at 500 °C between $\beta\text{Cu}_{2-x}\text{Se}$ and $\delta\text{Fe}_{1-x}\text{Se}$ yields $\beta\text{Fe}_{1.04}\text{Se}$.

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